

**UTILISATION OF AGRO WASTE FOR THE PRODUCTION OF PECTINASE  
BY MICROORGANISM**

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# CONTENTS

SL NO	TITLE	PAGE NO
1	INTRODUCTION	8 - 12
2	AIM AND OBJECTIVE	13
3	REVIEW OF LITERATURE	14 -35
4	MATERIALS AND METHODS	36 - 44
5	RESULTS AND DISCUSSION	45 - 58
6	SUMMARY AND CONCLUSION	59 - 60
7	REFERENCES	61 -65
8	APPENDIX	66 - 67

### LIST OF TABLE

SL No	TITLE	PAGE No
1	Pectinase producing Microorganism	28
2	Pectinase producing various substrates	29
3	Composition of basal medium	40
4	Composition of pectin	42
5	Composition of DNSA reagent	42
6	Pectinase assay	49
7	Pectinase activity on various temperature	52
8	Pectinase activity on various pH	53
9	Pectinase activity on various Incubation period	55
10	Pectinase activity on various nitrogen source	56

### LIST OF FIGURE

SL NO	TITLE	PAGE NO
1	Processing of Agro Waste	46
2	Bacillus Subtilis Culture In Nutrient Broth	47
3	Bacillus Subtilis Culture In Low Cost Substrate	48
4	Crude Enzyme	48
5	Pectinase Assay	50

### LIST OF GRAPH

SL NO	TITLE	PAGE
1	Pctinase assay	51
2	Effect of temperature	53
3	Effect of pH	54
4	Effect of incubation period	56
5	Effect of nitrogen source	58

## LIST OF ABBREVIATIONS

g	-	Gram
μl	-	microliter
%	-	percentage
mM	-	Millimolar
OD	-	Optical Density
pH	-	Potential of hydrogen
ml	-	Millilitre
DNSA	-	di -nitrosalicylic acid
Hr	-.	Hour
IP	-	Incubation period
NaoH	-	sodium hydroxide
M	-	molarity
PST	-	sodium potassium tartarate
NaHPO <sub>4</sub>	-	sodium hydrogen phosphate
DW	-	Distilled Water
KH <sub>2</sub> PO <sub>4</sub>	-	potassium dihydrogen phosphate
BWB	-	Boiling water bath
PG	-	Polygalacturonase
PE	-	Pectinesterase
PL	-	Pectatelyase
Ca	-	carbon
SSF	-	Solid state fermentation
N <sub>2</sub>	-	Nitrogen
°C	-	degree celsius
Rpm	-	Revolutions per minute
w/v	-	Weight per volume

## **ABSTRACT**

Pectinases are a big group of enzymes that break down pectic polysaccharides of plant tissues into simpler molecules like galacturonic pectin degrading enzymes were produced in solid state fermentation by *Bacillus subtilis* isolated from fruit and vegetable. Under optimized conditions, maximum Production of pectinase ( $3.190\mu/ml$ ) was recorded in the presence of orange peel with a moisture content Of 70% at 35 °C and pH 6.0 after 48 hrs of incubation, respectively. Pectinase yield was enhanced upon supplementation with ammonium sulphate and yeast extract. Thus, *B. subtilis* exploited for cost- effective production of pectinase using agro-residues.

**Keywords** pectinase. Solid state fermentation. Optimization

**Chapter 1**  
**INTRODUCTION**



Pectin is a heterogeneous structural polysaccharide present in primary cell wall and middle lamella of fruit and vegetable. It is a polymeric material having carbohydrate group esterified with methanol. It is an important component of plant cell wall. The highest concentration of pectin is present in middle lamella, where it acts as a cementing substance between adjacent cells. Pectin comprises of D – galacturonic acid occurred in - 1,4 chain, naturally esterified with methoxy group and natural sugars occupy the side chains.

Pectinase comprise a heterogeneous group of enzymes that catalyze the Breakdown of pectin containing substrates. Pectinase contribute to more than 25% of Global enzyme sales (Jayani et al. 2005). Pectinases are a group of enzymes involved in the depolymerisation of pectic polymers. Based on their mode of action, pectinases include polygalacturonase (PG), pectin esterase (PE), pectin lyase (PL), and pectate lyase (Ahlawat et al. 2009). Pectinase is involved in the hydrolysis of pectin containing a certain degree of esterified groups, while PG acts on unesterified polygalacturonic acid. Among all pectinases, pectin lyases are the only enzymes capable of depolymerizing highly esterified pectin into small molecules without prior action of other enzymes.

Microorganism enzymes are involved In the production of pectinase by using pectin as a carbon source.pectinsare degraded by several microorganisms that produce a variety of compounds and enzymes which are involved in several industrial applications.Many important bacteria,fungi and test are skillful at degrading pectin substances to produce pectinases.

Pectinases are used in fruit juice extraction and clarification, Wastewater treatment, vegetable oil extraction,tea and coffee fermentations, alcoholic beverages and in food industries (Jayaniet al .2005). In order to meet this high demand,it is important to produce pectin – depolymerizing enzymes at a large scale in a cost effective manner. Solid state fermentation (SSF) has gained more popularity in recent years in the production of many enzymes due to its lower operation costs and energy requirements, higher enzyme production and lower effluent generation as compared to submerged fermentation (SmF)(Pandey et al .1999;Couto and Sanroman 2006). The selection of substrate is important inSSF and depends on several factors, mainly the cost of availability.

Being both economical and eco-friendly, agro industrial residues are the prime choice of substrate for enzyme production. Orange peel was reported to contain 16.9% soluble sugar, 9.21% cellulose,10.5% hemicellulose and 42.5% pectin as the most important component ( Rivas et

al .2008). Orange peel a rich source of pectin containing about 20- 30% pectic substances (May 1990), is a good inducer of pectinase.

Orange trash, which makes up around 10%w/w of the weight of the orange fruit, is a by-product of the cutting of oranges into slices. Its current disposal raises significant economic and environmental concerns. Additionally, *B.subtilis* uses the orange waste as a low-cost production medium for pectinase.

The environmental concern about the disposal of solid waste has contributed to the increased interest in producing pectinaseutilising trash. Biotechnology's main goal is to better manage and utilise the enormous amounts of domestic, industrial, and agricultural waste. Organic materials are frequently crucial for both economic and environmental reasons.Numerous by- products of the agro-industry have poor economic value and are frequently dumped into waterways, where they contribute significantly to environmental damage. The use of biotechnology in these by removing a source of pollution, wastes can turn some of them into beneficial byproducts.

The importance of agro-industrial leftovers in bioprocesses helps to address pollution issues while also supplying new substrates. Many new opportunities for their use have emerged with the introduction of

biotechnological advancements, particularly in the field of enzyme and fermentation technology. The following are some biotechnology usage techniques for acceptable organic waste materials:

- Improve the food waste's quality so that it can be consumed by people.
- Recondition the food waste either before or after processing to be fed to fish, fowl, pigs, or other single-toothed animals that can use it right away.
- Give cattle or other ruminants food waste as a supplement.
- Production of compost, biogas (methane), and other products of fermentation
- Fuel, construction materials, and chemical extraction are examples of direct uses
- For the production of biological fuel or bioenergy.

The use of agricultural and industrial products and their wastes, including wheat bran, rice bran, ground nut peel, lemon peel, pineapple peel, papaya peel, banana peel and orange processing waste, for the production of pectinase using *B. subtilis* under SSF has attracted considerable interest in recent years. With the help of some bacteria, nearly all fruit waste produces a certain quantity of pectinase.

## **AIM AND OBJECTIVE**

### **AIM**

To optimize the production of pectinase from cheap agro waste

### **OBJECTIVES**

- Processing of agro waste
- Inoculum preparation for the fermentation
- Optimize the temperature,pH for the production of pectinase
- Perform enzyme assay
- Reduce the amount of agro waste

## **Chapter 2**

### **REVIEW OF LITERATURE**

Pectinases is responsible for catalysing the breakdown of pectic polymers found in plant cell walls. They are a member of the polysaccharidases family, which aids in the breakdown of Pectins come from many different plants and are also referred to as enzymes that break down or break up pectin (Prathyusha&Suneetha.(2011)In the current biotechnological era, pectinase are One of the forthcoming enzymes showing progressive Increase in their market. They maintained the average Annual growth rate of 2.86% from 27.6 million \$ in 2013 To 30.0 million \$ in 2016 and it is estimated that by 2021, The market size of the pectinase will reach 35.5 million \$ (Global Pectinase Market Research Report, 2017).

Pectinases are a heterogeneous group of related enzymes That hydrolyze the pectin substances, present mostly in Plants .Pectic enzymes are widely distributed in nature and Are produced by bacteria, yeast, fungi and plants (Babu and Bayer 2014). In plants, pectic enzymes are very important Since they play a role in elongation and cellular growth as Well as in fruit ripening (Jansirani et al. 2014).

Pectolytic Activity of microorganisms plays a significant role, firstly, In the pathogenesis of plants since these enzymes are the First to attack the tissue (Ovodov 2009). In addition, they Are also involved in the process of

symbiosis and the decay Of vegetable residues ( Hoondal et al. 2002). Thus by Breaking down pectin polymer for nutritional purposes, Microbial pectolytic enzymes play an important role in Nature (Yadav et al. 2009). These enzymes are inducible, produced only when needed and they contribute to the Natural carbon cycle (Hoondal et al. 2002).

### **Classification of Pectinases**

According to the cleavage site, pectinases are divided into Three groups: (1) hydrolases consisting of Polygalacturonase, PG (EC 3.2.1.15); (2) lyase/transeliminases comprising pectinlyase, PNL (EC 4.2.2.10), and Pectate lyase, PL (EC 4.2.2.2); (3) pectin esterase, PE (EC 3.1.1.11) (Yadav et al. 2009; Osborne 2004)

#### **1. Pectinesterase (PE)**

Pectin methy esterase or pectinesterase (EC 3.1.1.11) Catalyzes deesterification of the methoxyl group of pectin Forming pectic acid and methanol. The enzyme acts preferentially on a methyl ester group of galacturonate unit Next to a non-esterified galacturonate unit. It acts before Polygalacturonase and pectate lyases which need non esterified substrates (Kashyap et al. 2001).

#### **2. Polygalacturonase**



Polygalacturonases (PGases) are the pectinolytic enzymes that catalyse the hydrolytic cleavage of the polygalacturonicAcid chain with the introduction of water across the oxygen Bridge (Kashyap et al. 2001)

### **3. Pectatelyase**

Pectatelyase (PGL) cleaves glycosidic linkages preferentially on polygalacturonic acid forming unsaturated Product through transelemination reaction. PGL has an absolute requirement of  $\text{Ca}^{2+}$  ions. Hence it is strongly Inhibited by chelating agents as EDTA(Jayani, Saxena, and Gupta 2005a).

### **4. Pectin lyase**

Pectin lyase catalyzes the random cleavage of pectin, preferentially high esterified pectin, producing unsaturated methyloligogalacturonates through transelemination of glyosidic linkages. PLs do not have an absoluteRequirement of  $\text{Ca}^{2+}$  but they are stimulated by this and Other cations (Jayani, Saxena, and Gupta 2005b).

## **PECTIN**

Pectin, a polysaccharide, is composed primarily of Essentially linear polymers of D- galactopyranosyluronic acid units joined in  $\alpha$ -D (1->4) glycosidic Linkages; the polymer chains are esterified to Various degrees with methanol. This regular Structure is interrupted, however, with L-rhamnopyranosyl units and with side chains Containing other neutral

sugars. The polymer chains May also be partially acetylated. The most Important physical property of pectin is its ability To form spreadable gels. Gel formation results when The polymer chains interact over a portion of their Length to form a three-dimensional network. This aggregation of chains occurs through hydrogen Bonding, divalent cation crossbridging, and/or Hydrophobic interactions. Pectin is not a homopolysaccharide however and has rhamnopyranosyl residues inserted in the galactosyluronic backbone at 1 to 4% substitution. The other major feature of these rhamnogalacturonan-I (RG-I) chains are large substituted side chains. Between 20 and 80% of the rhamnopyranosyl residues are, depending on plant source and method of isolation, substituted at C-4 with neutral and acidic oligosaccharide side chains. The predominant side chains contain large linear and branched  $\alpha$ -L-arabinofuranosyl and /or  $\beta$ -D galactopyranosyl residues and their relative proportion and chain lengths may differ depending on plant source. Other rarer side chains are also present and generally shorter. The final and much more minor component of the backbone is rhamnogalacturonan-II (RG-II). This is not structurally related to RG-I since its backbone is composed of 1,4-linked  $\alpha$ -D-galactosyluronic residues like HG. At approximately 30 glycosyl residues long it has a non-saccharide and an octasaccharide side chain attached to C-2 of some of the backbone residues and two structurally

different disaccharides attached to C-3 of the backbone. RG-II is of interest as it occurs in relatively high amounts in wine and other fruit juices and it has been demonstrated that it binds heavy metals and has immunomodulating activities.

It is possible to separate essentially pure galacturonan fractions from other high molecular weight pectin fractions by degrading purified pectins specifically in the galacturonan backbone either chemically or enzymatically. It appears that there is an intramolecular distribution in which the neutral sugars are concentrated in blocks of more highly substituted rhamnogalacturonan regions ('hairy' regions) which are separated in the polymer by D-galactosyluronic-rich regions ('smooth' regions). These smooth regions can be up to 100 units in length.

### Physical Properties

Pectins are soluble in pure water, but they are insoluble in aqueous solutions in which they would gel at the same temperature if dissolved at a higher temperature. Monovalent cation (alkali metal) salts of pectin and pectic acids are usually soluble in water; di- and trivalent cation salts are weakly soluble or insoluble.

Although pectins are not employed as thickening agents, Pectin solutions exhibit the non-Newtonian, pseudoplastic behavior characteristic of most polysaccharides. As with solubility, the viscosity of a pectin solution is related to the molecular weight, DE, and concentration of the preparation and the pH and presence of counter ions in the solution.

These physical properties of pectins are a function of their structure which is that of a linear poly anion (polycarboxylate). As such, monovalent cation salts of pectins are highly ionized in solution, and the distribution of ionic charges along the molecule tends to keep it in an extended form by reason of coulombic repulsion.

#### Chemical properties

Dissolved pectins undergo deesterification and polymerization in aqueous systems. The pH of greatest stability is about 4. At pH values both above and below 4, deesterification and depolymerization occur concurrently, with the rate of deesterification being greater than the rate of depolymerization.

The presence of solutes, which lowers water activity, reduces the rates of both reactions.

There are several types of enzymes that act on pectin molecules. Those enzymes produced by the higher plants themselves play a significant role in

the processes resulting textural Changes in fruits and vegetables during ripening, storage, and Processing. Fungal enzyme preparations are used by the Fruit juice e industry to improve the clarity of juices and the Yield from processing.

### **Production of Microbial Pectinase**

Microorganisms are currently the primary source of industrial enzymes: 50% originate from fungi and yeast; 35% from bacteria, while the remaining 15% are either of plant origin. The microbial world has shown to be very heterogeneous in its ability to synthesize different types of pectolytic enzymes with different mechanisms of action and biochemical properties (Gummadi and Panda 2003). There were two fermentation techniques we can use for pectinases production, as many other enzymes. These techniques are Solid State Fermentation (SSF) SMF.

Solid state fermentation is defined as the cultivation of microorganisms on moist solid supports, either on inert carriers or on insoluble substrates that can be used as carbon and energy source. This process occurs in the absence or near absence of free water in the space between substrate particles. In this system, water is present in the solid substrate whose capacity for liquid retention varies with the type of material (Pandey, Soccol, Nigam, Soccol, et al. 2000). In

contrast, in submerged fermentation (SmF) the nutrients and microorganisms are both submerged in water.

Approximately 90% of all industrial enzymes are produced in SmF, frequently using specifically optimized, genetically manipulated microorganisms. In this respect SmF processing offers an insurmountable advantage over SSF. SSF has several advantages over SmF system such as higher concentration of products, less effluent generation, requirement for simple equipments etc (Pandey et al. 1999). The price of commercially available enzymes which are produced mostly by submerged fermentation is usually too high for agro-biotechnological applications. An alternative technique of enzyme production is solid state cultures (Kawano et al. 1999). Microbial production of pectinases has been studied during recent years (Kashyap et al. 2001). Pectinase production has been reported from bacteria including actinomycetes (Beg et al. 2000), yeast (Reid and Ricard 2000) and fungi. However, almost all the commercial preparations of pectinases are produced from fungal sources (Singh, Ramakrishna, and Appu Rao 1999). *Aspergillus niger* is the most commonly used fungal species for industrial production of pectinolytic enzymes (Gummadi and Panda 2003; Murad and Azzaz 2011; Jayani, Saxena, and Gupta 2005b).

Most extracellularly induced enzymes are known to be synthesized in higher quantities when inducers are present in the cultivation medium (Alkorta et al. 1998). The production of pectolytic enzymes using different sources and the effect of physical parameters such as temperature, aeration rate and type of fermentation were investigated and reported in literature (Naidu and Panda 1998). Pectolytic enzymes have been reported to be induced by several substances. In many cases pectin itself has been used. Many investigators had used complex media such as beet sugar, wheat bran, ground nut meal, citrus fruit peels etc (Hoondal et al. 2002).

Higher cost of the production is perhaps the major constraint in commercialization of new sources of enzymes. Though, using high yielding strains, optimal fermentation conditions and cheap raw materials as a carbon source can reduce the cost of enzyme production for subsequent applications in industrial processes (Murad and Azzaz 2011).

There are many studies that have been conducted related to the characterization of different microbial pectic enzymes concerning their mechanisms of action and biochemical properties. The optimal pHs that these enzymes may act range between 3.5-11, while the optimal temperatures vary between 40-75 °C (Gummadi and Panda 2003; Kashyap et al. 2001).

## Microbial sources

Different microorganisms are involved in the production of pectinase by using pectin as a carbon source. Pectins are degraded by several microorganisms that produce a variety of compounds and enzymes which are involved in several industrial applications. Many important bacteria, fungi and yeasts are skillful at degrading pectins substances to produce pectinases.

- Pectinolytic Fungi

Several fungal species can degrade pectic substances by producing pectinolytic enzymes. The most popular and more efficient fungi in the pectinase production are *Aspergillus Niger*, *Aspergillus awamori*, *Penicillium restrictum*, *Trichoderma viride*, *Mucor piriformis* and *Yarrowia lipolytica* have a great role in both submerged as well as solid-state fermentation for the production of various industrially important products. *Aspergillus Niger*, *Aspergillus oryzae* ,and *Penicillium expansum* are the types of fungi that are generally considered safe by the United States Food and Drugs Administration are put to use in the food industry [16]. Kumari et al. [49] isolate pectinase-producing strain *Penicilliumjanthinellum* from the soil and has been found to produce significant amounts of an extracellular pectinase subsequently characterized as exo- polygalacturonase. The different fungal strains from vegetable wastes and screened them for their



pectinolytic activity. Among them, *Tetracocco sporium* species was found to be good producers of pectinase and it showed a clearance zone of 20 mm pectinolytic activity around the colonies . Khairnar et al. studied the pectinase production of different strains of *Aspergillus Niger*. They observed the highest zone of clearance of pectin hydrolysis in *Aspergillus Niger* is 4.5 mm. Ten fungal isolates were isolated from municipal solid waste. Among them, a maximum zone of clearance of above 3.0 mm for pectinolytic activity was exhibited by *Penicillium chrysogenum* and *Aspergillus Niger* .

- Pectinolytic Yeasts

Kavuthodi and Sebastian [78] reported that *Saccharomyces fragilis*, *Saccharomyces thermantitonus*, *Torulopsis kefir*, *Candida pseudotropicalis* var, *lactosa*, and *Candida pseudotropicalis* are types of yeast that can degrade pectin substances in the pectinase production processes. The other report also indicates additional yeast species for pectinase production, these species include *Saccharomyces* species, *Cryptococcus* species, *Aureobasidium pullulans*, *Rhodotorula dairenensis*, *Kluyveromyces marxianus*, *Geotrichum klebahnii*, and *Wickerhamomyces anomalus*, [13, 79].

*Wickerhamomyces anomalus* one of the classification of species *Pichia anomala* produced pectinolytic enzymes in liquid medium containing glucose

and citrus pectin as carbon and energy sources. In the current studies, enzymes made by this wild yeast strain were characterized, and physicochemical properties of polygalacturonase were determined by the study of the influence of temperature and pH on its activity and stability to evaluate the application of the supernatant in the maceration of potato tissues [80]. The different investigations identified different yeast species and characterized in molecular method to the production of pectinolytic enzymes from grapes peel. Based on that identification, several species have a good potential to degrade pectin substance these species include: *Hanseniaspora* species, *Saccharomyces cerevisiae*, *Rhodotorula dairenensis*, *Candida zemplinina*, *Metschnikowia* species, *Aureobasidium pullulans*, and *Cryptococcus saitoi* [81].

- Pectinolytic Bacteria

*Erwinia* species, *Pseudomonas fluorescens*, *Bacillus*, *Pseudomonas*, and *Micrococcus* have a good potential to degrade pectin in the production of pectinase [20, 56, 78]. Other such as *Streptomyces* bacteria also has pectinolytic properties as reported by Ramirez-Tapias et al. [82]. *Bacillus licheniformis* has been reported as pectinolytic bacteria that were isolated from the rotten vegetable. The efficiency of *Bacillus licheniformis* to pectinase production was determined by the primary and secondary

screening methods. The primary screening was carried out by the potassium–iodide flooding method and the secondary screening was carried out by fermentation. The efficiency of *Bacillus licheniformis* on the pectinase activity was recorded as 341 U/ml [83]. A newly isolated *Brevibacillus borstelensis* reported good pectinase (pectin lyase) production and characterization. The enzyme activities of *Brevibacillus borstelensis* were reported as 5.25 U/ml [84]. Soil is collected from different villages of Guntur District (Duggirala and Burripalem) from a depth of 1–15 inches to isolate desired pectinase producing Bacteria.

Bacteria	<p>Bacillus sterothermophilus</p> <p>Bacillus cereus</p> <p>Bacillus mojavensis</p> <p>Erwinia app</p> <p>Bacillus subtilis</p> <p>Bacillustropicus</p> <p>Erwiniacarotovora</p> <p>Erwiniachrysanthemi</p>
Fungi	<p>Aspergillusoryzae</p> <p>Aspergillusflavus</p> <p>Aspergillussojae</p> <p>AspergillusNiger</p> <p>Penicilliumchrysogenum</p> <p>Trichodermaharzianum</p> <p>Schizophyllum commune</p>
Yeast	<p>Wickerhanomycesanomalus</p> <p>Saccharomyces cerevisiae</p> <p>Kluyveromycesmarxianus</p> <p>Filobasidiumcap suligenum</p>

Substrates used for pectinase enzyme production.

The pectinase enzyme production from various substrates have been identified and listed in the following table.

Substrate	Organism
Wheat bran	Bacillus subtilis
Date syrup	Bacillus subtilis
Agriculture waste	Bacillus pumilus
Cassava waste	Bacillus sp
Sugar beet pulp	Aspergillus Niger
Mango peel	Aspergillus foetidus
Lemon peel	Aspergillus oryzae
Banana peel	Aspergillus terreus
Wheat	Aspergillus awamori
Soy & wheat bran	Aspergillus Niger
Pectin	Bacillus subtilis
Citrus pectin	Bacillus sphaericus
Sucrose	Aspergillus Niger

Orange pomace	Aspergillus Niger
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### **Pectinase production by Bacillus species**

Pectic enzymes are of functional relevance in the retting process and evidence regarding pectinolytic properties of *Bacillus* spp. Was recorded years ago. Different species of the genus *Bacillus* have been reported to be retting agents and Active against pectic materials (Potter & McCoy, 1955). Nortje and Vaughn, 1953 tested the pectinolytic activity Of *B. subtilis* and *B. pumilus* in relation to the softening Of olives and pickles (Nortje & Vaughn, 1953). The first in Vitro fermentation studies of pectin and pectic acid was Reported in 1955 using *B. polymyxa* strain 30 (Potter & McCoy, 1955). Over the past few years,pectinolytic properties have been described in several *Bacillus* species .

It is evident from many research works that, among different bacterial isolates screened for pectinolytic properties *Bacillus* strains were selected as the most potent enzyme producers (Soares et al., 1999; Jayani et al., 2010; Rehman et al., 2012; Kavuthodi et al., 2015; Sohail& Latif, 2016). As mentioned in the introduction, alkaline pectinases have a wide variety of industrial applications and, bacteria mainly *Bacillus* spp are the chief producers. Apart from this fact, there are also some other

reasons for researchers to focus on pectinase from *Bacillus* spp. These include; (i) they produce all class of pectic enzymes, (ii) have short fermentation period for enzyme production, (iii) can produce enzymes very economically by using different agro-wastes as cheap substrates, (iv) fermentation can be attained by either SSF, Smf and (v) genetic information regarding pectinase genes of many *Bacillus* spp. Are available in various nucleotide sequence databases. Thus it supports successful cloning and expression of pectinase gene in other organisms.

### **Applications of Microbial Pectinases**

Application of enzymes in biotechnological process has expanded considerably in recent years. In food and related Industry, major importance was being attached to the use of Enzymes in upgrading quality, increasing yields of Extractive processes, product stabilization, and Improvement of flavor and by product utilization (Patil and Dayanand 2006). Pectinases or pectinolytic enzymes are Today one of the upcoming enzymes of the commercial sector.

It has been reported that microbial pectinases Account for 25% of the global food enzymes sales ((Jayani, Saxena, and Gupta 2005b). On the bases of their Applications, pectinases are mainly of two types: acidic Pectinases and alkaline pectinases (Jayani, et al., 2005; Murad and Azzaz 2011).

## **Acidic Pectinases**

Acidic pectic enzymes used in the fruit juice industries and Wine making often come from fungal sources, especially From *Aspergillusniger*(Kashyap et al. 2001). Potential applications of acidic pectinase are briefly described Below.

### **Fruit juice clarification/extraction**

Fruit juice clarification/extraction Is one among the Important applications of acidic pectinases Fruit juices contain colloids that may lead to fouling problem during filtration process and these colloids are basically polysaccharides such as pectin and starch (Rai et al. 2004). Pre-treatment of juices with pectinases is performed to lower the amount of pectin present and to decrease the viscosity of the juice, which in turn accelerates the subsequent filtration process. Also, it helps to increase the clarity of the juice.

### **Wine processing**

Wine processing industry also recognizes the importance of Acidic pectinases (Roldán et al. 2010), where the enzyme Can be applied at different stages. The addition of pectinases during crushing of the fruits increases the juice yield and Also accelerates the release of anthocyanins into the juice. Pectinase treatment at the pre-fermentation or fermentation Stage,



settles out suspended particles. After fermentation, Enzyme is added to the wine to increase its clarity and Filtration rate (Kashyap et al. 2001).

### **Tissue maceration**

Tissue maceration is another important application of acidic pectinases in which organized tissue is transformed into a suspension of intact cells and it is significant in the food industry as well as in the field of biotechnology. The process can be applied for the liquefaction and saccharification of biomass, isolation of protoplasts.

### **Alkaline pectinases**

Alkaline pectinases are mainly used in the degumming and Retting of fiber crops and pretreatment of pectic Wastewater from fruit juice industries. These enzymes come Mostly from bacterial sources (Kashyap et al.2001).In the Industrial sector, alkaline pectinases, mainly from *Bacillus* Spp. Are applied for the following purposes.

### **Paper and pulp industry**

During papermaking, pectinase can depolymerisepectins And subsequently lower the cationic demand of pectin Solutions and the filtrate from peroxide bleaching (Reid and Ricard 2000).

## **Animal feed**

Pectinases are used in the enzyme cocktail, used for the Production of animal feeds. This reduces the feed viscosity, Which increases absorption of nutrients, liberates nutrients, Either by hydrolysis of non-biodegradable fibers or by Liberating nutrients blocked by these fibers, and reduces the Amount of faeces(Jayani, Saxena, and Gupta 2005b).

## **Coffee and Tea Fermentation**

Fermentation of coffee using pectinolytic microorganisms is Done to remove the mucilage coat from the coffeebeans and To enhance the tea fermentation and foam forming property Of tea. Fungal pectinases are also used in the manufacture of Tea. Enzyme treatment accelerates tea fermentation, Although the enzyme dose must be adjusted carefully to Avoid damage to the tea leaf. Large-scale treatment of Coffee with commercial pectinases is costly and Uneconomical, inoculated waste mucilage is used as a Source of microbial pectin enzymes. The fermentation Liquid is washed, filtered and then sprayed on to the beans (Pandey, Soccol, Nigam, Brand, et al. 2000).

## **Textile processing and bio-scouring of cotton fibers**

Pectinases have been used in conjunction with amylases, lipases, cellulases and hemi-cellulases to remove sizing agents from cotton in a safe

and eco-friendly manner, replacing toxic caustic soda used for the purpose earlier. Bio-scouring is a novel process for removal of noncellulosic impurities from the fiber with specific enzymes. Pectinases have been used for this purpose without any negative side effect on cellulose (Hoondal et al. 2002).

### **Pectic waste water treatment**

The wastewater from the citrus-processing industry contains pectinaceous materials that are barely decomposed by microbes during the activated-sludge treatment have tried to develop a new wastewater treatment process by using an alkalophilic microorganism. Pretreatment of these wastewaters with pectinolytic enzymes facilitates removal of pectinaceous material and renders it suitable for decomposition by activated sludge treatment (Beg et al.2000).

### **Oil Extraction**

Citrus oils such as lemon oil can be extracted with pectinases. They destroy the emulsifying properties of pectin, which interferes with the collection of oils from citrus peel extracts(Mohnen 2008).

### **Purification of plant viruses**

Pectinases have also been reported to work on purification of viruses. But they are yet to be commercialized. When virus particle is restricted to phloem, to

release the virus from the tissues, alkaline pectinases and cellulases are used. This gives very pure preparations of the virus (Reid and Ricard 2000).

### **Retting and degumming of plant bast fiber**

Bast fibers are the soft fibers formed in groups outside the xylem, phloem or pericycle, e.g. Ramie and sun hemp. The fibers contain gum, which must be removed before its use for textile making. The chemical degumming treatment is polluting, toxic and non-biodegradable. Biotechnological degumming using pectinases in combination with xylanase presents an eco-friendly and economic alternative to the above problem (Kapoor et al. 2001). Pectinases have been used in retting of flax to separate the fibers and eliminate pectins (Hoondal et al. 2002).

## Chapter 3

### MATERIALS AND METHODS

## **Materials**

### **Instruments**

- Laminar air flow
- Spectrophotometer
- Centrifuge
- Incubator
- Rotary shaker

### **Chemicals**

- Nutrient broth
- Peptone
- Magnesium sulphate
- Sodium nitrate
- Pottasium dihydrogen
- Sodium hydrogen phosphate
- Glycine
- Sodium hydroxide
- Dinitrosalicylic acid
- Crystal phenol

- Sodiumsulphite
- PST
- Yeast extact
- Ammonium chloride
- Ammonium sulphate

## **APPARATUS**

- Test tubes
- Conical flask
- Micropipette
- Funnel
- Filter paper
- Glass rod
- Inoculation loop
- Test tube rack
- Measuring cylinder
- Glass beakers

Methods

## **PROCESSING OF AGRO- RESIDUE AS SUBSTRATE**

### **Collection of agro waste**

The waste from orange is chosen as the source of solid substrate for the production of pectinase. Orange fruits were collected from local fruit and vegetable market in kayamkulam . In a single batch, enough solid substrate were obtained for all the experiments.

### **Pretreatment of substrate**

Orange peel was removed and washed several times with tap water to remove water soluble compounds. Then it is dried in aluminium trays under sunlight for 1- 2 days. The substrate was dried until they were completely dehydrated. The dried orange peel were grounded to 2-3 mm particle size in a laboratory grinder and used for the production of pectinase. The powdered samples are gathered in sized vials and stored until they are needed.

### **MICROORGANISM**

*Bacillus subtilis* was obtained from stock culture in the laboratory.

### **Inoculum preparation**

The inoculum was prepared by transferring a loop full of *B. subtilis* culture into 50 ml autoclaved nutrient broth taken in 250 ml conical flask, which was then incubated at 37°C under agitation at 100 rpm on a rotary shaker for 18 hrs.

### **Preparation of nutrient broth**

- Weigh 0.65 g nutrient broth powder



- Take 49 ml distilled water in a measuring cylinder
- Mix the accurately weighed nutrient broth powder in DW
- Nutrient broth is prepared

Solid state fermentation for the production of pectinase

SSF was performed by Erlenmeyer flasks (250ml) .Each flask carrying 5 g of orange peel as substrate that is moistened with 70% distilled water before sterilization.The basal production medium in flasks was autoclaved,cooled and inoculated with 1.0ml of 18hr- old inoculum of *B.subtilis*.Flasks were then incubated at 37°C for 48 hrs in a bacterial incubator.

Basal medium Preparation (100 ml)

Component	Quantity g/ml
Peptone	0.1
KH <sub>2</sub> PO <sub>4</sub>	0.104
NaNO <sub>3</sub>	0.4
MgSO <sub>4</sub>	0.01
NaH <sub>2</sub> PO <sub>4</sub>	0.2
DW	93

## **Enzyme extraction**

Pectinase were extracted from the fermented substrate by constant shaking in an orbital shaker at 100 rpm for 20 min at 300c. The extract was squeezed through filter paper and centrifuged at 10000rpm for 5 min. The clear supernatant was collected as the crude enzyme.

## **Assay of pectinase**

Pectinase activity was assayed by measuring the amount of D-galacturonic acid liberated from pectin. The sample mixture contains 50  $\mu$ L of diluted enzyme and 450  $\mu$ L 0.5% pectin was incubated for 20 min at 300c and the end product was quantified by 3ml DNSA reagent and 1ml of PST. Then the sample place in boiling water bath for 5 min. After that cool the sample and measure the OD value at 510 nm.

## **REAGENT PREPARATION FOR ASSAY**

1. 0.5 % pectin

Component	Quantity g/ml
Pectin	0.005
NaoH	0.4
Glycine (0.1M)	0.7507
DW	100 ml

**DNSA REAGENT (20 ml)**

Component	Quantity g/ml
DNS	0.2
Crystal phenol	0.04
Sodium sulphite	0.01
NaoH	0.2
DW	20

- Dissolve by stirring 0.2 g DNS in 0.04 g crystalline phenol.
- Weigh 0.2 g NaOH and mix it with DW.
- Then weigh 0.01 g of sodium sulphite and add to DW
- Mix the DNS to the DW mixture

#### 40% Rochelle salt solution

PST – 2g

DW - 5ml

### **OPTIMIZATION**

#### **Effect of temperature**

The effect of cultivation of temperature were studied by incubating the three flask containing 5.0 g of orange peel at different temperature 25°C, 35°C, 40°C. The common procedure follows by enzyme extraction and pectinase assay.

#### **Effect of pH**

The common procedure follows by enzyme extraction and pectinase assay for pH 5, 6, & 9

#### **Effect of incubation period**

- Take 3 autoclaved conical flask
- Three conical flask contains 5 g of orange peel and moistened with 70% distilled water
- Add autoclaved basal medium to each flask
- Add inoculum to each flask within the LAF
- Kept the properly sealed flasks with time period of 12hrs ,24hrs, 48hrs,&72 hrs
- The common procedure follows by enzyme extraction and pectinase assay

### **Effect of nitrogen source**

The effect of nitrogen source for the production of pectinase by supplying different organic and inorganic source of nitrogen in solid media. This is carried out by four flask contains moistened orange peel ,basal medium and inoculum . Allow flask mixture to ferment. After fermentation, enzyme is extracted .The supernatant was collected as source of crude enzyme.Then it follows the pectinase assay.

## Chapter 4

### RESULTS AND DISCUSSION

## **1.POWERED SUBSTRATE**



Dried orange peel



**Powdered orange peel**

## 2. BACILLUS SUBTILIS CULTURED IN NUTRIENT BROTH





### 3.BACILLUS SUBTILIS CULTURED IN LOW COST SUBSTRATE



### CRUDE ENZYME FROM PECTINASE ASSAY



### Assay of pectinase activity

Pectinase activity was assayed by measuring the amount of D – galacturonic acid liberated from pectin. For enzyme assay, 1.0 ml of Bacillus culture was inoculated into flask containing basal medium. flask were incubated for 48 hrs. The fermented extract was filtered and centrifuged at 10000 rpm for 15 min .The supernatant is enzyme source. The enzyme unit was defined as the amount of enzyme that catalyzes  $\mu$ / mol of galacturonic acid per minute ( $\mu$ /mol min<sup>-1</sup>) under the assay conditions.

The enzyme activity was calculated thus the orange peel with 0.05 ml enzyme volume liberate 1.09 pectinase than orange peel with 0.1 ml enzyme volume liberate 3 .19 Therefore the Orange peel with 0.1ml of enzyme volume taken as standard for further experiments.

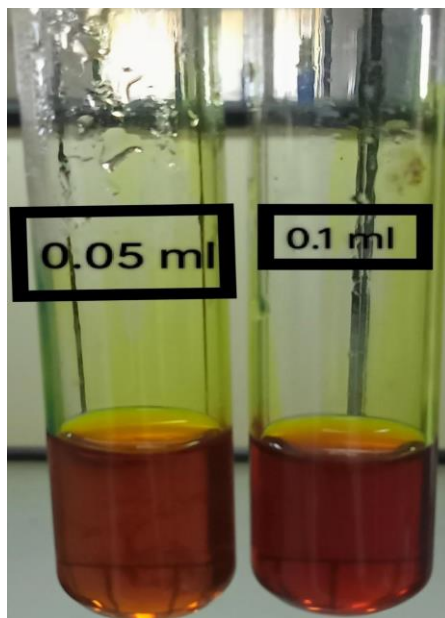
Enzyme activity expressed in

= Con of galacturonic acid x 1000x dilution factor

Molecular weight of galacturonic acid x incubation time

Substrate	Volume of enzyme ml	Concentration	OD at 510nm	Pectinase activity
Orange peel	0.05 ml	0.425	0.314	1.09
Orange peel	0.1 ml	0.620	0.352	3.19

## Assay of pectinase



Pectinase activity (0.05ml)

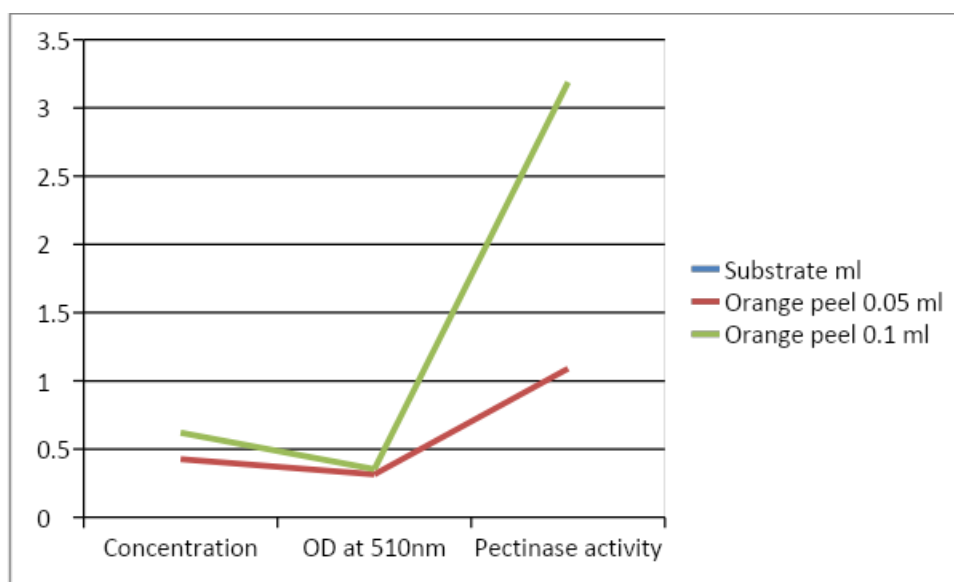
$$= \frac{0.425 \times 1000 \times 0.05}{194.139 \times 10}$$

$$= 1.09$$

Pectinase activity (0.1ml)

$$= \frac{0.620 \times 1000 \times 0.1}{194.139 \times 10}$$

$$= 3.19$$



## EFFECT OF TEMPERATURE

The orange peel with appropriate moisture and were inoculated and incubated at various temperatures. The maximum pectinase production attained at 35°C in fermentation technique. The succeeding studies were performed at 35°C.

SL no	Temperature	OD at 510 nm	Concentration	Pectinase activity
1	35	0.2	0.306	1.57
2	40	0.176	0.282	1.45
3	25	0.136	0.242	1.24

Temperature at 35°c

$$= \frac{0.306 \times 1000 \times 0.1}{194.139 \times 10}$$

$$= 1.57$$

Temperature at 40°c

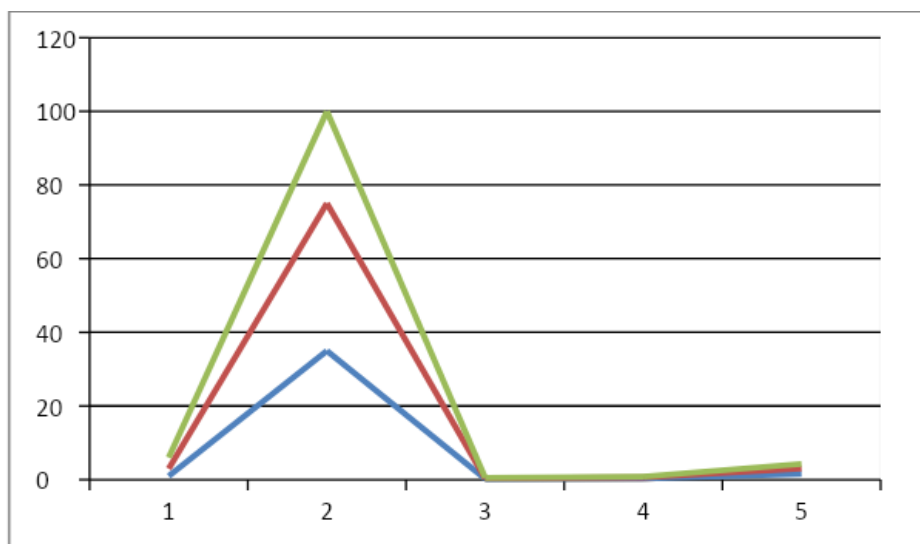
$$= \frac{0.282 \times 1000 \times 0.1}{194.139 \times 10}$$

$$= 1.45$$

Temperature at 25 °c

$$= \frac{0.242 \times 1000 \times .01}{194.139 \times 10}$$

$$= 1.24$$



### EFFECT OF pH

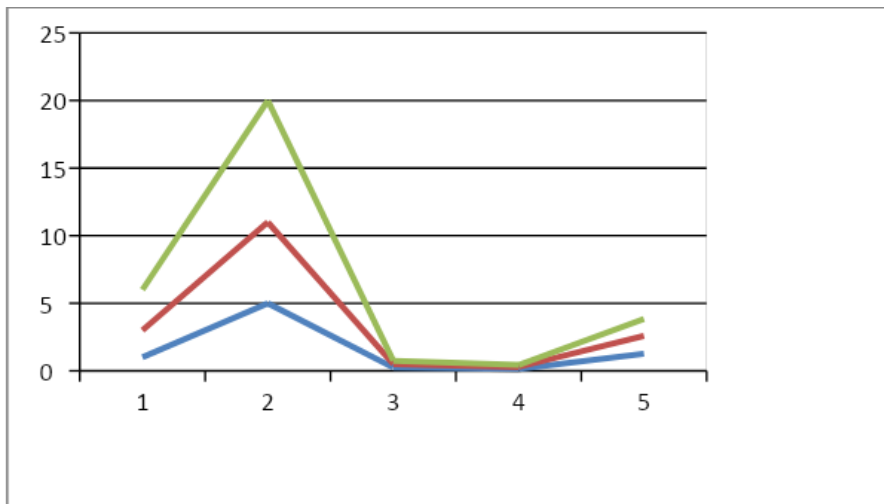
The effect of pectinase activity at different pH like pH 5 ,pH 6 , pH 9 is evaluated . The maximum amount of pectinase activity , Bacillus shows in pH 6 (1.31).In acidic pH , pectinase activity is high therefore the medium with pH 6 is selected for Further experiments.

SL No	PH	Concentration	OD at 510 nm	Pectinase activity
1	5	0.245	0.139	1.26
2	6	0.256	0.15	1.32
3	9	0.247	0.141	1.27

$$\begin{aligned}
 \text{pH 5} &= \frac{0.245 \times 1000 \times 0.1}{194.139 \times 10} \\
 &= 1.26
 \end{aligned}$$

$$\begin{aligned}\text{pH } 6 &= \frac{0.256 \times 1000 \times 0.1}{194.139 \times 10} \\ &= 1.31\end{aligned}$$

$$\begin{aligned}\text{pH } 9 &= \frac{0.247 \times 1000 \times 0.1}{194.139 \times 10} \\ &= 1.27\end{aligned}$$



### EFFECT OF INCUBATION PERIOD

The effect of incubation period, the orange peel as solid substrate mixed with basal medium inoculated and incubated were assayed for pectinase activity with different time period of incubation like 24,48,72,96 hrs. The highest pectinase enzyme production was achieved at 48 hrs of incubation (3.19). So 48 hrs of incubation period is taken for the further experiments.

SlNo	Incubation period	Concentration	OD at 510 nm	Pectinase activity
1	24	0.448	0.18	2.30
2	48	0.620	0.352	3.19
3	72	0.419	0.319	2.15
4	96	0.398	0.298	2.05

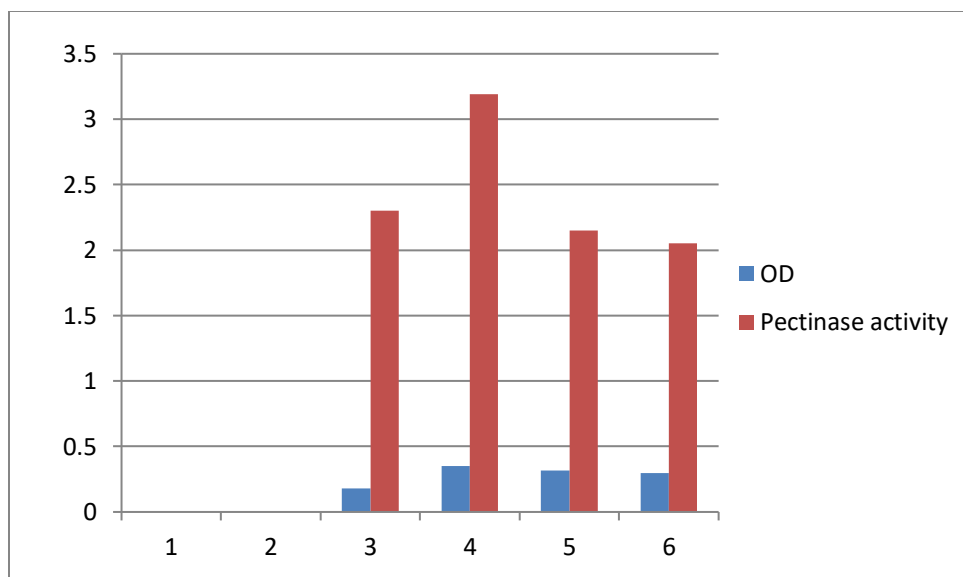
$$\begin{aligned}\text{IP (24)} &= \frac{0.448 \times 1000 \times 0.1}{194.138 \times 10} \\ &= 2.30\end{aligned}$$

$$\begin{aligned}\text{IP (48)} &= \frac{0.620 \times 1000 \times 0.1}{194.139 \times 10} \\ &= 3.19\end{aligned}$$

$$\begin{aligned}\text{IP (72)} &= \frac{0.419 \times 1000 \times 0.1}{194.139 \times 10} \\ &= 2.15\end{aligned}$$

$$\begin{aligned}\text{IP ( 96 )} &= \frac{0.398 \times 1000 \times 0.1}{194.139 \times 10} \\ &= 2.05\end{aligned}$$





## EFFECT OF NITROGEN SOURCE

The effect of nitrogen source on pectinase production the solid substrate supplemented with organic and inorganic sources of nitrogen 1 % ( w/v).The highest pectinase production attained at supplementing yeast extract (1.85) as organic nitrogen source. The highest pectinase production attained at supplementing Ammonium sulphate ( 1.83) as inorganic nitrogen source.so 1% yeast extract and 1% ammonium sulphate is the best source of nitrogen for pectinase production.

SINo	N source	Concentration	OD at 510nm	Pectinase activity
1	Yeast extract	0.361	0.255	1.85

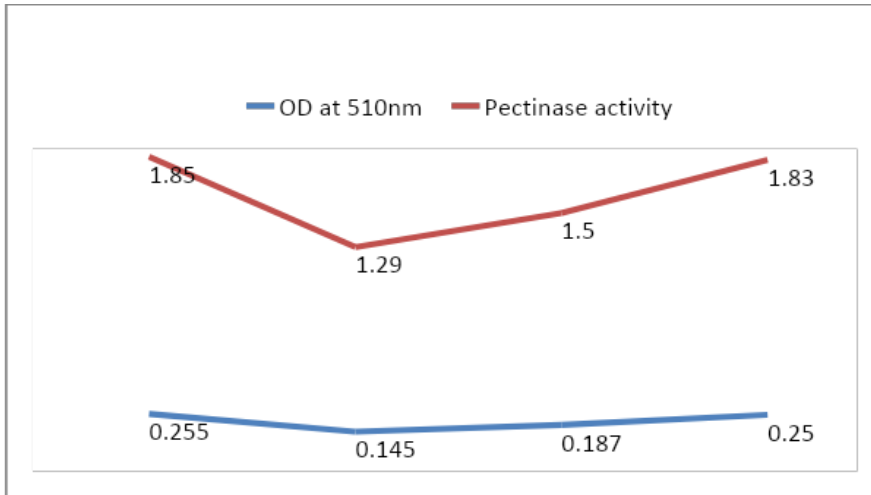
2	Peptone	0.251	0.145	1.29
3	NH <sub>4</sub> Cl	0.293	0.187	1.50
4	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	0.356	0.25	1.83

$$\begin{aligned}\text{Yeast extract} &= \frac{0.361 \times 1000 \times 0.1}{194.139 \times 10} \\ &= 1.85\end{aligned}$$

$$\begin{aligned}\text{Peptone} &= \frac{0.251 \times 1000 \times 0.1}{194.139 \times 10} \\ &= 1.29\end{aligned}$$

$$\begin{aligned}\text{NH}_4\text{Cl} &= \frac{0.293 \times 1000 \times 0.1}{194.139 \times 10} \\ &= 1.50\end{aligned}$$

$$\begin{aligned}(\text{NH}_4)\text{SO}_4 &= \frac{0.356 \times 1000 \times 0.1}{194.13 \times 10} \\ &= 1.83\end{aligned}$$



## Chapter 5

### SUMMARY AND CONCLUSION

Emerging pectinase applications highlight the value of selecting pectinase-producing microorganisms with unique characteristics, higher enzyme activity, and widespread production of these enzymes. Environmental factors like temperature, pH, incubation period, nitrogen source and the presence of substrates have an impact on microbes' ability to synthesise extracellular enzymes.

In this study, parameters that affect the pectinase production have been standardized and diligent optimization steps were carried out to make the production of pectinase enzyme to be cost effective and commercially viable.

Since, to meet the growing industrial demands for pectinase, it is necessary to improve yield without increasing the cost of production. Thus, in this study the biotechnological capacities of agricultural wastes are considered for economical production of pectinase.

This study aimed evaluating the production and activity of pectinase from agro waste by using *Bacillus subtilis*. The results from the study showing that we can produce pectinolytic enzyme using agro waste as substrate, bacillus as Microorganism, allow this set up for solid state fermentation. After fermentation the extracted enzyme under assayed condition we can obtain the amount of pectinase liberated. The parametric optimization of temperature, pH, incubation period, nitrogen source enhanced the pectinase production. The production of pectinase was enhanced more than a 6-fold in solid state fermentation.

The potential of agricultural wastes for the production of pectinase using solid state fermentation is highlighted in this study. This result conveys the very economized production of pectinase. This is an efficient method for cost reduction in the production of enzymes.

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## APPENDICES

### • Basal medium Preparation (100 ml)

It is prepared by dissolving basal medium in 100ml distilled water

Component	Quantity g/ml

Peptone	0.1
KH <sub>2</sub> PO <sub>4</sub>	0.104
NaNO <sub>3</sub>	0.4
MgSO <sub>4</sub>	0.01
NaH <sub>2</sub> PO <sub>4</sub>	0.2
DW	93

- **0.5 % pectin preparation**

It is prepared by dissolving 0.5% pectin in 100ml distilled water

Component	Quantity g/ml
Pectin	0.005
NaOH	0.4
Glycine (0.1M)	0.7507
DW	100 ml

**DNSA REAGENT PREPARATION (20 ml)**

Component	Quantity g/ml
DNS	0.2

Crystal phenol	0.04
Sodium sulphite	0.01
NaoH	0.2
DW	20

- Dissolve by stirring 0.2 g DNS in 0.04 g crystalline phenol.
- Weigh 0.2 g NaoH and mix it with DW.
- Then weigh 0.01 g of sodium sulphite and add to DW
- Mix the DNS to the DW mixture

#### **40% ROCHELLE SALT SOLUTION** **PREPARATION**

It is prepared by dissolving 2g PST in 5  
ml of distilled water

PST – 2g

DW - 5ml